# Department of Textile Engineering

B. Sc. Engineering 1st Year Backlog Examination, 2018

### Math-1221

(Mathematics-II)

Time: 3 Hours

Total Marks: 210

- N.B.: i) Answer any THREE questions from each section in separate scripts.
  - ii) Figures in the right margin indicate full marks.
  - iii) Assume reasonable data if missing any.

### **SECTION-A**

- 1(a) Find the rotational angle of the axes to remove the xy term from the 11 equation  $3x^2 + 2xy + 3y^2 = 1$ . Hence find the transformed equation.
- 1(b) Find the transformed coordinates of (-2,1) with respect to 3x 4y + 1 = 0 and 12 4x + 3y 1 = 0 as the x-axis and y-axis respectively. Also find the distance from the point (-2,1) to the point of intersection of these two axes.
- 1(c) Identify the conic represented by the equation: 12  $x^2 5xy + y^2 + 8x 20y + 15 = 0$ , and also find its standard form.
- 2(a) Find the rectangular and spherical polar coordinates of the point A, whose cylindrical coordinate is  $(3, \frac{2\pi}{3}, 4)$ .
- 2(b) Find the equation of the plane through the point (2, -1, -4) and perpendicular to the 11 planes 3x + 4y 5z + 6 = 0 and x 2y + 2z + 1 = 0.
- 2(c) Find the distance from a point P(2,3,-4) to the plane x+y+6z-2=0 measured 12 parallel to the line  $\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-3}{-2}$ .
- 3(a) Test whether or not the straight lines  $\frac{x-3}{-3} = \frac{y-8}{1} = \frac{z-3}{-1}$  and  $\frac{x+3}{-3} = \frac{y+7}{2} = \frac{z-6}{4}$  are 15 coplanar. If they are non-coplanar, then find the shortest distance between them.
- 3(b) A line in 3-D space makes angle 90° with x and z axes. Find its direction cosines.
- 3(c) Find the standard form of the equation of a line-3x + 2y - 3z - 4 = 0 = 4x + y - z + 3.
- 4(a) Define great circle. Find the equation of the sphere, for which the circle  $x^2 + y^2 + 13$  $z^2 + 7y - 2z + 2 = 0$ , 2x + 3y + 4z = 8 is a great circle.
- 4(b) Define right circular cone. Find the semi-vertical angle of the right circular 10 cone,  $z^2 2x^2 2y^2 = 0$ . Also find its axis.
- 4(c) Find the coordinates of the point where the line joining the points (3, -4, -5) and 12 (2, -3, 1) cuts the plane 2x + y + z 12 = 0.

5(a) Formulate a differential equation from the relation:

$$y = c_1 e^{-2x} \cos 3x + c_2 e^{-2x} \sin 3x$$

Where  $C_1$  and  $C_2$  are arbitrary constants. Also write the order, degree and state whether it's linear or nonlinear.

- 5(b) Solve the differential equation 9yy' + 4x = 0 and sketch two integral curve from the solution.
- 5(c) Find the general solution of  $\frac{du}{dt} = -k[u T_0 A \sin t]$ .
- 6(a) Find an integrating factor for the equation  $(3xy + y^2) + (x^2 + xy)y' = 0$  and then 10 solve the equation.
- 6(b) Solve  $\frac{dy}{dx} = \frac{x+2y-3}{2x+y-3}$ .
- 6(c) A body at a temperature  $40^{\circ}F$  is placed outside where the temperature is  $90^{\circ}F$ . If after 13 5 minutes the temperature of the body is  $50^{\circ}F$ , find the time required by the body to reach  $60^{\circ}F$ .
- 7(a) Solve  $(D^2 + 1)y \approx \tan x$  by using the method of variation of parameter.
- 7(b) Solve  $y'' + 4y = x^2 + \sin 2x + e^{-x}$ .
- 7(c) Find the work done in moving an object along a straight line from (3,2,-1) to 10 (5,-1,7) in a force field given by  $\underline{F} = 4\underline{i} 3\underline{j} + 2\underline{k}$ .
- 8(a) Show that the vector field represented by  $\vec{F} = (2xz^3 + 6y)\hat{\imath} + (6x 2yz)\hat{\jmath} + (3x^2z^2 y^2)\hat{k}$ is conservative and hence find its scalar potential.
- 8(b) If  $\vec{F} = 4xz\hat{\imath} y^2\hat{\jmath} + yz\hat{k}$ , evaluate  $\iint \vec{F} \cdot \hat{n} \, ds$

Where S is the surface of the cube bounded by x = 0, x = 1; y = 0, y = 1; z = 0, z = 1.

8(c) Find the acute angle between the surfaces  $x^2 + y^2 + z^2 = 9$  and  $z = x^2 + y^2 - 3$  at 11 the point (2, -1, 2).

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# Department of Textile Engineering

## B. Sc. Engineering 1st Year Backlog Examination, 2018

### Ph-1221

(Physics)

Time: 3 Hours

Total Marks: 210

- N.B.: i) Answer any THREE questions from each section in separate scripts.
  - ii) Figures in the right margin indicate full marks.
  - iii) Assume reasonable data if missing any.

## **SECTION-A**

1(a) What is interference of light? Discuss interference of light analytically and obtain the 12 conditions of maximum and minimum intensities. 1(b) Discuss Young's double slit experiment and hence show that, the spacing between two 13 consecutive bright and dark fringes are the same. In a Newton's rings experiment the diameter of the 15th ring was found to be 0.590 cm 10 1(c)and that of the 5th ring was 0.336 cm. If the radius of the plano-convex lens is 100 cm, calculate the wavelength of the light used. Discuss the Fraunhofer diffraction at a single slit. Draw the intensity distribution for 2(a) the diffraction pattern. What is polarization of light? Explain Brewster's Law and show that, when light is 2(b) incident on a transparent substance at the polarizing angle, the reflected and refracted rays will be at right angle. 2(c) A 20 cm long tube containing sugar solution rotates the plane of polarization by 11°. If 10 the specific rotation of sugar is 66°, calculate the strength of the solution. Derive an expression for magnification of compound microscope. 12 What is Compton effect? Explain and derive an expression for Compton shift  $\Delta\lambda$  on the -133(b) basis of quantum theory. The threshold frequency of photoelectric emission in copper is  $1.1 \times 10^5$  Hz. Find the 3(c) maximum energy of photoelectrons (in eV) when light of frequency 1.5x10<sup>15</sup> Hz is directed on a copper surface. Discuss uncertainty principle and show that electron cannot stay in the nucleus with the 4(a) help of uncertainty principle. Derive an expression for energy of an electron of hydrogen atom. 4(b) 13 Calculate the limiting value of wavelength for Lyman series. 10

- 5(a) Explain the concept of Miller indices. How are they calculated? How the orientation of a plane is specified by Miller indices?
- 5(b) Define Primitive cell and Atomic Packing fraction. Show that, for FCC structure, 10 atomic packing fraction is 74%.
- 5(c) Show that for a cubic lattice, if (hkl) are corresponding Miller indices, then interplanar 10 spacing can be written as,

$$d_{hkl} = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

- 6(a) Describe the formation of energy band in solids. Explain how it helps to classify the 10 materials into metals, insulators and semiconductors.
- 6(b) Show that according to Einstein model, lattic heat capacity can be written as-

$$C_{v} = 3R \left(\frac{\theta_{E}}{T}\right)^{2} \frac{e^{\frac{\theta_{E}}{T}}}{\left(e^{\frac{\theta_{E}}{T}} - 1\right)^{2}}$$

Hence, at low temperature, compare the result with experimental observation.

- 6(c) Show that the ratio between thermal and electrical conductivity is proportional to 10 absolute temperature.
- 7(a) What is Hall effect? Show that, Hall coefficient  $R_H$  can be written as -

$$R_H = \frac{1}{ne}$$

- 7(b) Show that average kinetic energy of a free electron is  $\frac{3}{5}E_f$ , where  $E_f$  is Fermi energy 12 and average speed is  $\frac{3}{4}v_f$ , where  $v_f$  is the velocity of Fermi surface.
- 7(c) Calculate the inter collision time at room temperature and drift velocity in a field of  $10 \ 100 \ Vm^{-1}$  in sodium, whose conductivity is  $2.16 \times 10^7 \Omega^{-1} m^{-1}$ .
- 8(a) What is LASER? Discuss the properties of LASER and give some applications of 13 Laser.
- 8(b) Discuss the following terms briefly
  - i) Population inversion
  - ii) Stimulated emission
  - iii) Spontaneous emission
- 8(c) Discuss the fundamental theory of a chemical LASER.

# Department of Textile Engineering

B. Sc. Engineering 1st Year Backlog Examination, 2018

### Math-1121

(Mathematics-I)

Time: 3 Hours Total Marks: 210

- N.B.: i) Answer any THREE questions from each section in separate scripts.
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## **SECTION-A**

1(a) A function f(x) is defined as follows:

$$f(x) = \begin{cases} -2x - 2 \text{ for } x < -3\\ 4 \text{ for } -3 \le x < 1\\ 2x + 2 \text{ for } x \ge 1 \end{cases}$$

Discuss the continuity and differentiability of f(x) at x = 1.

- 1(b) At a certain instance the side of a square is a 3 ft long and increasing at a rate of 2 12 ft/minute. How fast is the area increasing at that instance?
- 1(c) Find the *nth* derivative of- $y = \frac{1}{x^2 3x + 2} + 3\cos 3x$
- 2(a) State Leibnitz's theorem. 12

  If  $y = (\sin^{-1} x)^2$ , then find  $y_{n+2}$  by using Leibnitz's theorem.
- 2(b) If u = f(y z, z x, x y), then determine the value of  $\frac{\partial u}{\partial x} + \frac{\partial u}{\partial y} + \frac{\partial u}{\partial z}$ .
- 2(c) State mean value theorem and verify it for  $f(x) = x^2 + 3x + 2$  in [1,2].
- 3(a) State Rolle's theorem. Is Rolle's theorem applicable for the function  $f(x) = \frac{1}{4-x^2} \text{ in } [-1,1]?$
- 3(b) Evaluate sin 31° using the value of sin 30° by Taylor's series.
- 3(c) Find the equation of tangent and normal at the point (1, -1) to the curve  $x^3 + xy^2 3x^2 + 4x + 5y + 2 = 0$ .
- 4(a) Determine the radius of curvature at the origin of the curve  $y x = x^2 + 2xy + y^2$ . 12
- 4(b) Determine all the asymptotes of the curve  $x^3 2x^2y + xy^2 + x^2 xy + 2 = 0$ . 13
- 4(c) Evaluate  $\lim_{x\to 0} \left(\frac{\tan x}{x}\right)^{\frac{1}{x}}$

5(a)	Calculate $\int \frac{dx}{(x^2+2)\sqrt{x^2+3}}$ .	11
5(b)	Evaluate $\int \frac{2\sin x + 3\cos x}{4\cos x + 5\sin x} dx$	12
5(c)	Calculate $\int \sin^{-1}(\sqrt{\frac{x}{2+x}}) dx$	12
6(a)	Obtain a reduction formula for $\int \sin^2 x  dx$ and hence evaluate $\int \sin^5 x  dx$ .	11
6(b)		
0(0)	Define Beta function and Gamma function. Establish the relation between Beta function and Gamma function.	14
6(c)	Determine the value of-	10
	$\lim_{n\to\infty} \left[ \left( 1 + \frac{1}{n^2} \right)^{\frac{2}{n^2}} \left( 1 + \frac{2^2}{n^2} \right)^{\frac{4}{n^2}} \left( 1 + \frac{3^2}{n^2} \right)^{\frac{6}{n^2}} \dots \dots \left( 1 + \frac{n^2}{n^2} \right)^{\frac{2}{n}} \right]$	
7(a)	Evaluate $\int_0^1 \frac{\log(1+x)}{1+x^2} dx$	12
7(b)	Evaluate $\int_0^{\pi} \frac{x \tan x}{\sec x + \cos x} dx$	12
7(c)	Evaluate $\int_0^{\frac{\pi}{2}} \frac{dx}{1+\sqrt{\cot x}}$	11
8(a)	Test whether the set of vectors $\{(2,-1,4),(3,6,2),(2,10,-4)\}$ are linearly independent	11
	or not.	
8(b)	Find the inverse of the matrix A, if exists, where-	12
	$A = \begin{bmatrix} 1 & 2 & 3 \\ 2 & 3 & 0 \\ 0 & 1 & 2 \end{bmatrix}$ By elementary row transformations.	

8(c) Solve the following system of linear equations

2x + 2y + 3z = 14

2x + 3y + 4z = 20

3x + 4y + 6z = 33

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# Department of Textile Engineering B. Sc. Engineering 1st Year Backlog Examination, 2018

### TE-1123

## (Polymer Engineering)

Time: 3 Hours Total Marks: 210

N.B.: i) Answer any THREE questions from each section in separate scripts. ii) Figures in the right margin indicate full marks.

iii) Assume reasonable data if missing any.

1(a)	Define: i) Monomer, ii) Homopolymer, and iii) Repeating unit.	06
1(b)	Describe the criteria of fiber forming polymer.	10
1(c)	Write down the monomer and repeat unit of the following polymers:	09
	i) Nylon, ii) Polyestyrene, and iii) Poly ethene	-
1(d)	Write down the importance of polymer in textiles.	10
		-
2(a)	What is polymerization?	05
2(b)	Describe the free radical polymerization with examples.	12
2(c)	Show the relation between DP and extent of reaction.	10
2(d)	What is Nylon 6:6? How this is formed?	08
		٠,
3(a)	Define polymer degradation. Write down the changes happened due to polymer	08
``.	degradation.	Ų.
3(b)	How polymer degradation can be controlled?	07
3(c)	Write short notes on the followings:	07
	i) Anti-oxidants, and ii) Hydrolytic	
3(d)	Describe photo-degradation process of polymer.	13
4(a)	What are $T_m$ and $T_g$ ? Write down the relation between $T_m$ and $T_g$ .	10
4(b)	Discuss the factors that influence the glass transition temperature of polymer.	10
4(c)	What is softening point temperature? Briefly describe the measuring process of	15

5(a)	Mention the names of different polymerization techniques. What are the reasons for	08
	choosing different polymerization techniques?	
5(b)	Describe the Bulk polymerization with advantages and disadvantages.	12
5(c)	State the different steps of emulsion polymerization.	10
5(d)	Write short note on suspension polymerization.	05
6(a)	What is meant by liquid crystal phase?	05
6(b)	Discuss the adjacent reentry chain-folded model for smooth and rough surface with sketch.	10
6(c)	Why some polymers are highly crystalline and some are highly amorphous? Explain.	12
6(d)	Write short note on spherulite structure.	08
7(a)	Why tensile strength, impact strength, and chemical resistivity of a polymer increases with the increase of molar mass?	09
7(b)	Differentiate between number average and weight average molecular weight.	06
7(c)	What are the different types of polymer molecular weight? Give the equation for each type.	10
7(d)	What are the properties dependent on polymer molecular weight? Sketch the graph of molecular weight versus polymer properties.	10
8(a)	Write down the functions of different zones of extruder in polymer processing machine.	08
8(b)	Describe the injection molding process of thermoplastic polymer with proper sketch.	12
8(c)	Derive the following expression for rate of polymerization in chain growth polymerization:	15
	$rate = K[I]^{\frac{1}{2}}[M]$	
	Where [I] is initiator concentration, [M] is monomer concentration and K is constant.	

# Department of Textile Engineering

B. Sc. Engineering 1st Year Backlog Examination, 2018

(Fundamentals of Mechanical Engineering)

Total Marks: 210 Time: 3 Hours

N.B.: i) Answer any THREE questions from each section in separate scripts.

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:	SECTION-A	-
1(a)	Define i) System, ii) Cycle, and iii) Process.	06
1(b)	Derive the relation between $C_P$ and $C_V$ .	06
1(c)	What is an adiabatic process? Derive an expression for the work done during adiabatic expansion of an ideal gas.	10
1(d)	A container contains $0.5  m^3$ of gas at a pressure of 2.5 bar and $170^{\circ} C$ . It is	13
	compressed adiabatically to a pressure of 10.0 bar. Determine the work required. Take	
	$C_P = 1.06 \text{ Kj/kg.K} \text{ and } C_V = 0.731 \text{ Kj/kg.K}.$	
2(a) 2(b)	State and explain Zeroth law of thermodynamics.  What is PMM1? Is it possible or not? Explain.	06
2(c)	What are meant by reversible and irreversible process? Explain the phenomena about	08
,	the cause of irreversibility and condition for reversibility.	
2(d)	Describe the working principle of a four stroke diesel engine.	12
3(a)	Classify IC engine. Write down the advantages and disadvantages of IC engine.	08
3(b)	Describe the working principle of four stroke diesel engine.	10
3(c)	Differentiate between petrol engine and diesel engine.	07
3(d)	During the test on single cylinder oil engine, working on the four stroke cycle and	10
	fitted with a rope brake, the following readings are taken: Effective diameter of brake	
	wheel= 630 mm; Dead load on brake= 200 N; Spring balance reading= 30 N;	
	Speed= 450 rpm; Mean effective pressure= 7.7 bar; Calorific value of	
	Oil = 42000  Kj/kg.	
	Calculate brake power, indicated power, and mechanical efficiency.	
4(a)	Draw the $P-V$ and $T-S$ diagram of i) Bryton cycle, ii) Dual cycle, and iii) Diesel	09
	cycle.	
4(b)	What is gas turbine? Classify gas turbine with necessary explanation.	09
4(c)	What is pump? Write down the classification of pump. Describe the working principle	11
	of a reciprocating pump.	
4(d)	Write short note on i) Compressor, and ii) Blower.	06

5(a)	What is meant by steam generating unit? Write down the differences between fire tube boilers and water tube boilers.	10
5(b)	Describe the working principle of "La-Mont" boiler with necessary sketches.	10
5(c)	Write down the functions of the following components in a steam boiler:	15
` '	i) Safety valve, ii) Water level indicator, iii) Feed pump, iv) Air preheater, and v)	·:
	Economizer.	
6(a)	Define i) Refrigerant, ii) Tonne of refrigeration, and iii) C.O.P.	06
6 <u>(</u> b)	State the properties of a good refrigerant. Establish how an actual cycle differs from a theoretical vapour compression cycle.	10
6(c)	Show that $(COP)_{HP} = (COP)_R + 1$	07
6(d)	Describe the working principle of Vapour Absorption Refrigeration System with neat	12
	sketch.	
7(a)	Define the following terms:	09
	i) Specific humidity, ii) Wet bulb temperature, and iii) Dew point temperature.	
7(b)	Describe the working principle of "Summer air conditioning" system with neat sketch.	12
7(c)	Define comfort. What are the factors which affect comfort air conditioning?	07
7(d)	What is condensation? Distinguish between film wise condensation and drop wise condensation.	07
8(a)	Define pool boiling. Describe the pool boiling phenomena of water with mentioning its	12
9/5)	regime with a clear sketch.  What is black body? State Stefen-Boltzman's Law and Wein's displacement law.	08
8(b)		
8(c)	Consider slab of thickness L. The boundary surface at $X = 0$ and $X = L$ are maintained at constant but different temperature of $T_1$ and $T_2$ respectively. There is no energy	15
	generation in the solid and the thermal conductivity K is constant. Develop an expression for the temperature distribution $T_{(x)}$ in the slab and the thermal resistance of	
	the slab for the heat flow through an area A.	

# KHULNA UNIVERSITY OF ENGINEERING & TECHNOLOGY Department of Textile Engineering B. Sc. Engineering 1st Year Backlog Examination, 2018

### TE-1221

(Textile Fibers)

Time: 3 Hours

Total Marks: 210

- N.B.: i) Answer any THREE questions from each section in separate scripts.

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  iii) Assume reasonable data if missing any.

1(a)	What is textile fiber? Classify the textile fibers with examples.	12
1(b)	Discuss the morphological diagram of cotton fiber mentioning the chemical composition.	.;. <b>15</b>
1(c)	What is ginning? Differentiate between lint and linters.	.08
2(a)	State the grading system for jute and cotton.	15
2(b)	Describe the physical and chemical properties of cotton fiber.	15
2(c)	Why jute is called bast fiber?	05
3(a)	Write down the chemical composition of jute fibers. How and why retting is done?	<sub>,:</sub> 10
3(b)	Discuss the micro structure of wool fiber with neat sketch.	,10
3(c)	State a procedure of distinguishing wool fiber from silk fiber.	05
3(d)	Discuss the chemical composition and favorable properties of Coir and banana fiber.	10
4(a)	What is Sericulture? Describe the production stages of silk filament.	15
4(b)	Write short notes on: i) Degumming of silk, and ii) Carbonizing.	08
4(c)	Mention the chemical composition and physical properties of silk	12

5(a)	Describe the characteristics of fiber forming polymers.	10
5(b)	Explain the spinning methods of man-made fibers with their merits and demerits.	20
5(c)	Mention the spinning systems for the following fibers:	
٠.	i) Carbon, ii) Nylon, iii) Polyester, iv) Viscose-rayon, and v) Glass fiber.	05
6(a)	Describe the manufacturing process of polyester fibers.	. 12
6(b)	State the chemical properties of polyester fibers.	10
6(c)	Mention the MR% and MC% for the following fibers:	
	i) Cotton, ii) Jute, iii) Nylon, iv) Polyester, and v) PALF.	05
6(d)	Differentiate between Nylon 6 and Nylon 66.	· 08
7(a)	Write short notes on: i) Carbon fiber, ii) Kevlar fiber, and iii) Glass fiber.	12
7(b)	What are acrylic and modacrylic? Write down the physical and chemical properties acrylic fiber.	of 12
7(c)	Describe the manufacturing process of polypropylene fiber.	11
8(a)	State the various steps involved in viscose process.	12
<sup>11</sup> 8(b)	What is elastomeric fiber? Show the flowchart of production of spandex fiber.	15
8(c)	Show the polymerization reaction of acetate and triacetate.	08

# Department of Textile Engineering B. Sc. Engineering 1st Year Backlog Examination, 2018

### Ch-1121

(Chemistry-I)

Time: 3 Hours

Total Marks: 210

- N.B.: i) Answer any THREE questions from each section in separate scripts.
  - ii) Figures in the right margin indicate full marks.
  - iii) Assume reasonable data if missing any.

# **SECTION-A**

	,	
1(a)	What is electrophoresis? How can you prove that colloid particles are electrically	11
	charged?	
1(b)	Define coagulation and precipitation. In between $MgCl_2$ and $KCl$ , which one is the	12
	better coagulant and why?	
1(c)	Explain dialysis and electrodialysis methods for the purification of sols.	12
2(a)	Explain photosensitized reaction with relevant example.	07
2(b)	State Stark-Einstein law of photochemical equivalence. Explain the term	12
2(0)	"Quantum Yield" in a photochemical reaction.	
2()		08
2(c)	The quantum efficiency for the hydrogen and chlorine reaction is very high, why?	
2(d)	Calculate the energy (in joule) associated with one photon of wavelength 8000 Å.	08
	$(h = 6.62 \times 10^{-27} \text{ erg-sec})$	
-		i
3(a)	What is bleaching agent? Compare the bleaching mechanism of NaClO <sub>2</sub> and HOCl.	. 12
3(b)	Explain with an example why pH of a buffer solution does not change significantly on	13
	small addition of acids or bases.	
3(c)	Discuss the Ostwald's theory of acid-base indicator.	10
	and the state of the	
4(a)	What is bond order? Predict the bond order and magnetic properties of $O_2$ and $H_2$	12
. ( <del>u</del> )	molecule.	
471-5		12
4(b)	How does molecular orbital theory differ from valence bond theory? Explain with	12
	suitable example.	,
4(0)	"N is distance but Ne is managamic" - Fynlain by MOT	- 11

5(a)	What is equivalent conductance? Show graphically the variation of equivalent conductance against concentration for $NaCl$ and $CH_3COOH$ . Explain the nature of	12
	these curves.	
5(b)	State Kohlrausch's law of independent migration of ions.	د 12
5(c)	Define transport number and establish the relation $t_+ + t = 1$ .	11
6(a)	Explain that the fully exposed metal surface is more beneficial than the pa vally	10
	exposed metal surface from the corrosion point of view.	
6(b)	Explain why the presence of acid increases but base decreases the rate of under-water corrosion.	10
6(c)	"Bolts and nuts are made of the same metal in practice"- why?	07
6(d)	Define the following terms with examples:	08
	i) Hydrolysis, ii) Pyrolysis	
		÷. Z
7(a)	What are the colligative properties and why are they so called?	07
7(b)	Deduce from Raoult's law an expression relating the molecular mass of a solute with	10
	the lowering of vapour pressure.	
7(c)	State and explain the Vant Hoff's laws of osmotic pressure.	10
7(d)	Show that the elevation of boiling point is directly proportional to the lowering of vapour pressure.	08
8(a)	Write down the IUPAC name of the following complexes:	06
	i) $[CO(NH_3)_4Cl_2]^+$ , ii) $K_4[Fe(CN)_6]$	
8(b)	Indicate the primary and secondary valencies of the central metal in the following complex compounds: i) $[CO(NH_3)_6]Cl_3$ , ii) $[CO(NH_3)_4Cl_2]Cl$	10
8(c)	Apply VB theory to explain hybridization, shape, and magnetic behavior of the	12
	complex $[Fe(CN)_6]^{4-}$	
8(d)	What is meant by "Effective Atomic Number"? Calculate the EAN in the following	07
	complex	

 $[Fe(CN)_6]^{3}$ 

# Department of Textile Engineering B. Sc. Engineering 1st Year Backlog Examination, 2018

### Ch-1221

(Chemistry-II)

Time: 3 Hours Total Marks: 210

N.B.: i) Answer any THREE questions from each section in separate scripts. ii) Figures in the right margin indicate full marks.

iii) Assume reasonable data if missing any.

# **SECTION-A**

1(a)	Define functional group isomerism and geometrical isomerism.	. 10
1(b)	What is plane polarized light? Explain.	07
1(c)	Explain the terms enantiomers and meso compound.	10
1(d)	What is racemic mixture? Explain.	08
•		
2(a)	Define staggered and eclipsed conformations giving examples.	. 08
2(b)	Outline the stereochemistry of cyclohexane giving energy diagram.	10
2(c)	Discuss the stereochemistry of $SN_1$ and $SN_2$ reactions.	10
2(d)	Predict about R and S configuration of the compound $CH_3C^*ClFBr$ .	07
3(a)	Define aromaticity and ring current.	10
3(b)	Discuss the mechanism of Friedel-Craft's reaction.	10
3(c)	Mention two important methods of synthesis of benzene.	<b>08</b>
3(d)	What is meant by free radical and carbocation? Explain.	07
4(a)	How can you distinguish between a ketone and an aldehyde?	08
4(b)	Alcohols and amines are highly soluble in water. Why?	09
4(c)	Methylamine is stronger base than ammonia. Why?	. 08
4(d)	Synthesize 2° amine and 3° alcohol.	10

	SECTION-B	: .
5(a)	What are carbohydrates? How are they classified?	12
5(b)	What is mutarotation? Explain with the help of an example.	13
5(c)	What are epimers? Differentiate between anomers and epimers.	10
6(a)	What are polysaccharides? Write the composition of starch.	10
6(b)	Deduce the structure of amylopectin by end group analysis.	13
6(c)	What happens when cellulose is treated with:	12
	i) $HNO_3 + H_2SO_4$ , ii) Acetic acid + Acetic anhydride.	
7(a)	Discuss the chemical properties of dyes and pigments.	10
7(b)	Discuss the modern theory of color.	12
7(c)	What are dye intermediates? Briefly discuss the application of dye intermediates.	13
8(2)	Explain the terms is a electric point and pentide linkage	10
8(a)	Explain the terms iso-electric point and peptide linkage.	10
8(b)	Most of the α-amino acids are optically active. Why?	07
8(c)	What are amino acids? Mention their importance.	08
8(d)	How can you detect C-terminal and N-terminal residue of proteins?	10

# Department of Textile Engineering B. Sc. Engineering 1st Year Backlog Examination, 2018

### Hum-1221

(Business and Communicative English)

Time: 3 Hours

Total Marks: 210

- N.B.: i) Answer any THREE questions from each section in separate scripts.
  - ii) Figures in the right margin indicate full marks.
  - iii) Assume reasonable data if missing any.

1(a)	Frame W	/H questions from the following answer:	1 4
	i)	Dr. Smith is the author of this book.	
	ii)	Jack's plan succeeded at last.	
	iii)	You must go home now.	
	ĭv)	I depend on authentic books for data collection.	
	v)	I want to know his address.	
	vi)	He is moving towards south.	
	vii)	The manager was late due to traffic jam.	
1(b)	Make sen	tences using the following words as directed-	12
	Pen (as	verb); Pen (as adjective); Money (as adjective); Place (as verb);	
	While (as	conjunction); What (as pronoun).	`
1(c)	Make sen	tences using the following phrases and idioms:	69
	A piece of	f cake, add insult to injury, all at once, black sheep, for good, in time	
2(a)	Correct th	ne following sentences:	14
	i)	I am the younger in the family.	
	ii)	He is comparatively better today.	
	iii)	Most of the apples is rotten.	
	iv)	One must do her work.	
	v)	Cattles are grazing in the field.	
	vi)	He rests in evening.	
	vii)	He gets a good pay.	
2(b)	Transform	n the following sentences:	12
	i)	Her work is better than yours. (Positive)	
	ii)	The police tried all plans. (Negative)	
	iii)	Their glory can never fade. (Interrogative)	
	iv)	I met a strange man. (Complex)	
	v)	He danced as if he were an expert dancer. (Simple)	
	vi)	In spite of my annoyance, I kept quiet. (Compound)	.*
2(c)	Define pre	esent participle, infinitive and gerund with example.	09

3(a)	Complete the following sentences with subordinate clauses:						
	i)	My mother taught me to read					
	ii) Mina went on reading						
iii) The journey took longer than							
	iv) The passage is so difficult						
vi) Go to bed if							
	vii)	Shella had to wait an hour					
3(b)	Make one	synonym and one antonym of each of the following words and use them in	12				
	sentences.						
	Adverse;	Bonafide; Cordial.					
3(c)	Supply a s	suitable word to fill in the blanks of the following:	09				
	i)	Hearing the news, he hurried home.					
	ii)	at home, he met his parents.					
	iii)-	walking along the road, he met me.	, .				
	iv)	He was the talkative lady.					
	v)	The man is to die.					
	vi)	The police men are for the criminals.					
4(a)	Make sen	tences using the following modals as directed-	14				
	i)	May (To express guess about the future)					
	ii)	May (To express guess about the present)					
	iii) <sup>*</sup>	Could (To express past ability)					
	iv)	Could (To express polite request)					
	v)	Be to (To express arrangement)					
	vi)	Be to (To express command)					
٠	vii)	Had better (To express performance)	•				
4(b)		ne following notions in sentences:	12				
	Worry, an	ger, surprise, disappointment, sympathy, revenge					
4(c)	Identify th	ne parts of speech of the underlined words of the following sentences.	09				
	i)	He runs <u>fast</u> .					
	ii)	I fast every monday.					
	-						
	iii)	Respect your <u>better</u> .					
	iii) iv)	Respect your <u>better</u> .  I know English <u>better</u> than you.					
		•					

5(a)	Read the	naccade a	nd answer	the c	mestions
J(a)	Read the	passage a	iid aiiswci	uic c	jucanona.

Advertising has become a very specialized activity in modern times. In the business world of today, supply is usually greater than demand. There is a great competition between different manufacturers of the same king of product to persuade customers to buy their own particular brand. They always have to remind the consumer of the name and qualities of their product. They do this by advertising. The manufacturer advertises in the newspaper and on poster. He employs attractive sales girls to distribute samples of it. He organizes competitions, with prizes for the winners. He often advertises on the screens of local cinemas. Advertisements are also broadcasted on television and radios. Manufacturers often spend large sum of money on advertisements. We buy a particular product because we think it is the best as the advertisements said so. Some people never pause to ask themselves if the advertisements are telling the truth.

- i) How many kinds of advertisements are mentioned in the passage? What are they?
- ii) Why do manufacturers spend so much money on advertising?
- iii) Which do you think is the most effective advertisement?
- iv) Do you think we buy goods because they are advised? Give reasons for your answer.
- 5(b) Make a précis of the above passage.

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6(a) Write a list paragraph on a good student.

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6(b) Amplify the idea contained in the following statement:

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- Borrowed garments never fit well.
- 7(a) Write a report on 21<sup>st</sup> February you observed in your campus.

15 20

- 7(b) Write a letter to the editor of a newspaper about student indiscipline and suggesting remedies.
- 8(a) Write a free composition on any one of the followings:

35

- a) Student politics: Its merits and demerits.
- b) Religious festivals and their values.